# PLANT ANTIMUTAGENS, 6.<sup>1</sup> INTRICATIN AND INTRICATINOL, NEW ANTIMUTAGENIC HOMOISOFLAVONOIDS FROM HOFFMAN0SSEGGIA INTRICATA

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ABSTRACT.—Intricatin [1] and intricatinol [2] are new homoisoflavonoids isolated from a desert plant Hoffmanosseggia intricata, which was collected in Baja California, Mexico. The structures of the compounds were elucidated using a variety of spectroscopic techniques. The structure of 1 was shown to be 7,4'-dimethoxy-8-hydroxyhomoisoflavone, and 2 was shown to be 4'-methoxy-7,8-dihydroxyhomoisoflavone. Both compounds 1 and 2 displayed activity in the inhibition of the mutagenicity of 2AN toward Salmonella typhimurium (T98). Compound 2 was much more active than 1 in the case of the inhibition of the mutagenicity of AAF toward S. typhimurium (T98) and the inhibition of EMS towards S. typhimurium (T100). Compounds 1 and 2 are the first examples of antimutagenic activity in homoisoflavones.

Two new closely related homoisoflavonoids, intricatin [1] and intricatinol [2], have been isolated from the roots of *Hoffmanosseggia intricata* Brandegee (Fabaceae), guided by the inhibition of the mutagenicity of 2-aminoanthracene (2AN) toward *Salmonella typhimurium* (T-98) by crude and purified extracts and concentrates (1-5). Homoisoflavonoids, with one expection, have previously been reported to occur only in plants which are members of the Liliaceae family (6-10). This paper reports the characterization and antimutagenic or toxic properties of new homoisoflavones 1 and 2 isolated from a nonrelated family, Fabaceae.



## **EXPERIMENTAL**

ANTIMUTAGENIC TESTING.—The procedures for determining the inhibition of the mutagenicity of 2AN toward cultures of *S. typhimurium* (T-98) by crude plant extracts or purified products and toxicity determinations have been described in previous papers (1–5). Assays for antimutagenicity of pure compounds toward 2AN and acetylaminofluorene (AAF), which require metabolic activation, and ethyl methanesulfonate (EMS), which does not, have also been described (1–5).

GENERAL ISOLATION AND CHARACTERIZATION METHODS.—Melting points were determined on a Kofler hotstage microscope and are uncorrected. <sup>1</sup>H- and <sup>13</sup>C-nmr spectra were obtained with a Bruker WM250 spectrometer using TMS as an internal standard and CDCl<sub>3</sub> as a solvent. High resolution mass spectra were obtained with an AEI MS-902 instrument. Uv spectra were obtained in MeOH with a Varian 2290-UV-VIS spectrometer and ir spectra with Perkin-Elmer 467 Grating spectrometer. Standard chromatography was carried out on Si gel E. Merck 230–40 mesh or Baker Flash chromatography Si gel using, in general, CH<sub>2</sub>Cl<sub>2</sub> as eluent with a gradient of 0.5–10.0% MeOH. For tlc determinations precoated Si gel plates were utilized; normal phase EM precoated Si gel 60, F254, usual solvent 10% MeOH in

<sup>&</sup>lt;sup>1</sup>For Part 5 of this series, see Manikumar et al. (5).

 $CH_2Cl_2$ ; reversed-phase, Baker precoated Si gel  $C_{18}$ -F plates, usual developer 5–10%  $H_2O$  in MeOH. Exposure of plates to iodine vapor was used as a general detection agent; alternatively, spraying with phosphomolybdate reagent followed by heating was utilized.

PLANT MATERIALS.—Roots of *H. intricata* were collected in March 1980, in Baja California, Mexico, by Richard Spjut, Plant Medicinal Resources Laboratory ARS. A voucher specimen is deposited in the herbarium of the National Arboretum, ARS.

EXTRACTION AND ISOLATION.—Roots (0.2 kg) were extracted with hot 95% EtOH. The EtOH extract was concentrated and the residue partitioned between H<sub>2</sub>O and CHCl<sub>3</sub>. After concentration of the CHCl<sub>3</sub> fraction, the crude residue (8.7 g) was tested for inhibition of 2AN mutagenicity at 0.6 mg/plate. Two separate assays showed inhibition of 38 and 50%. The CHCl<sub>3</sub> residue was nontoxic at 0.6 mg/plate. The CHCl<sub>3</sub> residue (4.0 g) was chromatographed on 50 g of Baker Si gel and was eluted initially with 0.5% MeOH in CH<sub>2</sub>Cl<sub>2</sub> (fractions 1–50). Fractions 30–50 yielded inhibitory material (0.3 g). Further elution with 2% MeOH in CH<sub>2</sub>Cl<sub>2</sub> yielded inhibitory material (0.06 g) in fractions 76–120. The material from fractions 30–50 was crystallized from aqueous MeOH, yielding 0.044 g of yellow crystals of a new compound designated intricatin [1] (7,4'-dimethoxy-8-hydroxyhomoisoflavone), 62% inhibition of 2AN at 0.3 mg/plate, nontoxic at 0.6 and 0.3 mg/plate. The more polar fraction was crystallized from aqueous MeOH, yielding 0.12 g of yellow needles of a new compound designated intricatinol [2] (4'-methoxy-7,8dihydroxyhomoisoflavone), toxic at 0.6 mg, nontoxic at 0.3 mg/plate, 87% inhibition of 2AN at 0.3 mg/ plate.

Intricatin [1].—Mp 157–159°, uv  $\lambda$  max (MeOH) nm (log  $\epsilon$ ) 347 (4.41), 237 sh (4.27), (MeOH + NaOH) 393 (4.48), 324 (4.20), 269 (3.99); ir  $\nu$  max (CHCl<sub>3</sub>) 3510 (OH), 1660 (C = O), 1600 (aromatic), 1470, 1300, 1175, 840, 820 cm<sup>-1</sup>; <sup>1</sup>H nmr (Me<sub>2</sub>CO-d<sub>6</sub>)  $\delta$  3.82 (3H, s, OMe), 3.88 (3H, s, OMe), 5.48 (2H, d, J = 2 Hz, H-2), 6.65 (1H, d, J = 8.8 Hz, H-6), 7.06 (2H, d, J = 8.8 Hz, H-3', -5'), 7.43 (2H, d, J = 8.8 Hz, H-2', -6'), 7.61 (1H, d, J = 8.8 Hz, H-5), 7.72 (1H, broad triplet, H-9); hrms m/z (rel. int. %) [M]<sup>+</sup> 312.0990 (C<sub>18</sub>H<sub>16</sub>O<sub>5</sub> = 312.0998) (100), 297 (15), 167 (55), 146 (70), 121 (35).

Intricatinol [2].—Mp 196–198°, uv  $\lambda$  max (MeOH) nm (log  $\epsilon$ ) 351 (4.44), 234 sh (4.24), (MeOH + NaOH) 396 (4.33), 320 (4.17), 286 (3.93); ir  $\nu$  max (KBr) 3375, 1660, 1600, 1270, 840, 815 cm<sup>-1</sup>; <sup>1</sup>H-nmr (DMSO-*d*<sub>6</sub>) 2 broad OH peaks at  $\delta$  10.0 and  $\delta$  8.7 exchanged with D<sub>2</sub>O, 3.83 (3H, s, OMe), 5.39 (2H, d, J = 2 Hz, H-2), 6.57 (1H, d, J = 8.5 Hz, H-6), 7.06 (2H, d, J = 8.5 Hz, H-3', -5'), 7.27 (1H, d, J = 8.5 Hz, H-5), 7.42 (2H, d, J = 8.5 Hz, H-2', -6'), 7.65 (1H, broad s, H-9); hrms *m*/z (rel. int. %) [M]<sup>+</sup> 298.0845 (C<sub>17</sub>H<sub>14</sub>O<sub>5</sub> = 298.0841) (30), 153 (65), 152 (40), 146 (100).

## **RESULTS AND DISCUSSION**

The uv and ir spectra of intricatin [1] and intricatinol [2] are similar to those described for the structurally similar homoisoflavones bonducellin (9) and 8-methoxybonducellin (10). Hrms of 1 gave a molecular formula of  $C_{18}H_{16}O_5$  in accord with the assigned structure. Ions at m/z 312 [M]<sup>+</sup>, 297 [M - 15]<sup>+</sup>, and 281 [M - 31]<sup>+</sup> are characteristic. Of greater diagnostic value, however, was a strong fragment peak at m/z167 indicative of retro-Diels-Alder rearrangement with one OH and one OMe moiety in ring A (11). Hence, the remaining OMe group must be in ring B.

The <sup>1</sup>H-nmr spectrum at 250 MHz (Me<sub>2</sub>CO-d<sub>6</sub>) revealed the presence of two methoxy groups with signals at  $\delta$  3.88 and  $\delta$  3.82. One methoxy group was assigned to the 4' position in agreement with literature (6–10) and the other to ring A based on the mass spectral fragmentation pattern (11). A perturbed AA', BB' system had two protons centered at  $\delta$  7.43 and two at  $\delta$  7.06 which were assigned to H-2', -6' and H-3', -5', respectively, similar to the data reported for bonducellin (9,10). The spectra showed two ortho coupled protons (J = 8.5 Hz) at  $\delta$  7.61 and  $\delta$  6.65 assignable to H-5 and H-6, respectively. A proton due to H-9 appeared at  $\delta$  7.72 as a broad triplet as a result of coupling with the two methylenic protons at position 2, which appeared as a doublet at  $\delta$  5.48 (J = 2 Hz). The position of the H-9 proton is indicative of a trans double bond at this position (6). In CDCl<sub>3</sub>, a D<sub>2</sub>O exchangeable proton (8-OH) appeared at  $\delta$  6.3. The location of the methoxy groups at C-7 and C-4' was deduced from the C<sub>6</sub>H<sub>6</sub>-induced shifts of methoxy groups. Only methoxy groups ortho to an aromatic proton are shifted upfield by 0.5 ppm in C<sub>6</sub>D<sub>6</sub> compared to CDCl<sub>3</sub> (12). We found both





methoxy groups at C-7 and C-4' shifted upfield by 0.5-0.6 ppm, confirming the structure of **1** as 7,4'-dimethoxy-8-hydroxyhomoisoflavone.

The uv and ir spectra of 2 closely resembled those of 1 and 8-methoxybonducellin (10). Hrms showed that the molecular formula of 2 was  $C_{17}H_{14}O_5$ . A strong fragment of m/z 153 was indicative of a retro-Diels-Alder rearrangement with two OH moieties in ring A (11).

The nmr spectrum (250 MHz) of **2** in DMSO- $d_6$  revealed the presence of one methoxy group at  $\delta$  3.83 assigned to the C-4' position and two D<sub>2</sub>O exchangeable hydroxyl groups at  $\delta$  10.0 and 8.7. A perturbed AA', BB' system had two protons centered at  $\delta$  7.42 and two at  $\delta$  7.06 which were assigned to the 2', 6' and 3', 5' positions, respectively. There were two ortho coupled protons (J = 8.5 Hz) centered at  $\delta$  7.27 and  $\delta$  6.57. These signals corresponded to H-5 and H-6, respectively. The two aliphatic protons at position 2 appeared as a narrow doublet at  $\delta$  5.39 (J = 2 Hz), and H-9 appeared at  $\delta$  7.65. The relationship of H-2 and H-9 protons was confirmed by long range 2D-proton correlation spectra (COSYLR) (Figure 1). Hence the structure of **2** is 7,8-dihydroxy-4'-methoxyhomosioflavone.

Intricatin [1] was nontoxic at 0.6 and 0.3 mg/plate; intricatinol [2] was toxic at 0.6 mg but nontoxic at 0.3 mg/plate. Table 1 shows the inhibition by 1 and 2 of the mutagenicity toward of *S. typhimurium* of 2AN, AAF, and EMS. The first two require metabolic activation but the latter is a direct acting mutagen which is known to alkylate the various DNA bases (13). Compound 1 was active in the inhibition of the mutagenicity of 2AN toward *S. typhimurium* (T98) but did not inhibit the mutagenicity of AAF, *S. typhimurium* (T98), or EMS, *S. typhimurium* (T100). The lack of dose response for 1 shown in Table 1 at doses of  $300-37.5 \mu g$  is anomalous; usually a dose response is observed over this dose range (cf. Table 1, compound 2). Compound 1 was inactive in inhibition of the mutagenicity of 2AN, AAF, and EMS (Table 1). The greater potency of 2 may be due to the presence of a dihydroxy moiety in ring A. In a previous study with flavonoids (2), we found that hydroxylated flavonoids were generally more active than their methylated analogues.

	Mutagen*								
Compound	2AN <sup>a,b</sup> Positive Control 2.5 μg/plate 1794 Colonies/plate <sup>c</sup>			AAF <sup>4,b</sup> Positive Control 25 μg/plate 1131 Colonies/plate <sup>c</sup>			EMS <sup>a</sup> Positive Control 750 µg/plate 2046 Colonies/plate <sup>c</sup>		
	Dose µg/plate	Colonies <sup>c</sup> #/plate	Inhibition <sup>d</sup> %	Dose µg/plate	Colonies <sup>c</sup> #/plate	Inhibition %	Dose µg/plate	Colonies <sup>c</sup> #/plate	Inhibition %
1	300 150 75	673 279	62 84 92	300 150 75	1047 1341 1162	7 0	300 150 75	1126 1491 1966	45 27 4
2	37.5 300 150 75	543 226 773	70 87 57 36	37.5 300 150 75	1035 335 518 590	8 70 54 48	37.5 300 150 75	1854 526 363	9 74 82 74
	37.5	1654	8	37.5	840	26	37.5	1438	30

 TABLE 1.
 Inhibition of Mutagenicity of Various Mutagens by Intricatin [1] and Intricatinol (2].

<sup>\*</sup>2AN = 2-aminoanthracene; AAF = acetylaminofluorene; EMS = ethyl methanesulfonate.

<sup>b</sup>Requires metabolic activation.

<sup>c</sup>Average of 3 plates.

<sup>d</sup>% Inhibition =  $\frac{\text{Colonies in positive control} - \text{Colonies in experimental plate}}{\times 100}$ 

Colonies in positive control

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